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C: When $n = 3$, $l = 0, 1,$
or 2 . The highest value
for l in this case is 2 ,
which corresponds to
the d subshell.

Although the $3d$ block
appears to be part of
the fourth period, it still
has the principal
quantum number $n =$
 3 . In general, the
subshells within an
energy shell increase

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in energy as follows: $s < p < d < f$ (although there is no 3f subshell).

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